

Comparing the Membrane-Interaction Profiles of Two Antiviral Peptides: Insights into Structure–Function Relationship

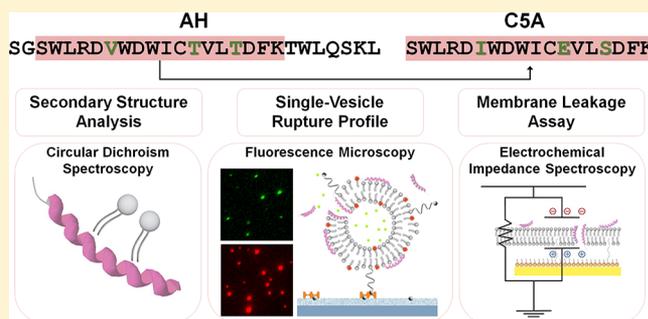
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Supporting Information

ABSTRACT: In recent years, certain amphipathic, α -helical peptides have been discovered that inhibit medically important enveloped viruses by disrupting the lipid membrane surrounding individual virus particles. Interestingly, only a small subset of amphipathic, α -helical peptides demonstrate inhibitory activity, and there is broad interest in understanding how the structures of these peptides contribute to functional activity against lipid membranes. To address this question, herein, we employed multiple surface-sensitive measurement techniques along with computational simulations in order to investigate how AH and C5A peptides, two of the most biologically active peptides in this class, interact with model lipid membranes while gaining insight into membrane-induced peptide conformational changes. Circular dichroism spectroscopy experiments revealed that both AH and C5A peptides undergo pronounced coil-to-helix transitions in the presence of lipid membrane environments, and the C5A conformational change was the largest. Time-lapsed fluorescence microscopy measurements were conducted to monitor the interaction of peptides with arrays of tethered, individual lipid vesicles and showed that C5A potentially lyses lipid vesicles indiscriminate of vesicle size at peptide concentrations as low as 10 nM whereas AH peptide preferentially lyses lipid vesicles with high membrane curvature and is less potent than C5A. These findings were complemented by electrochemical impedance spectroscopy measurements on a tethered lipid bilayer membrane platform, which indicated that C5A solubilizes lipid membranes in a manner that is distinct from how AH disrupts lipid membranes via pore formation. Computational simulations supported that the distinct membrane-interaction profiles arise from different helical folding patterns, whereby AH monomers predominantly exist as two shorter helices with a hinge in-between and C5A monomers form a single helix. Taken together, our findings demonstrate that membrane-active antiviral peptides can exhibit distinct membrane-interaction profiles that confer different degrees of targeting selectivity, and the corresponding structural insights will be useful for peptide engineering applications.



INTRODUCTION

Membrane-active peptides are important anti-infective agents that target infectious pathogens by destabilizing phospholipid membrane coatings.^{1–3} The classical description of anti-infective peptides typically focuses on cationic antimicrobial peptides that selectively destabilize membranes enriched in negatively charged lipids (e.g., bacterial cell membranes).^{4,5} In recent years, there has been increased attention on another exciting class of anti-infective peptides termed antiviral peptides, which can directly impair the lipid membrane surrounding enveloped virus particles.⁶ This antiviral approach has several compelling features, including broad-spectrum inhibitory potential against multiple classes of enveloped viruses, and there is a high barrier to the emergence of drug-resistant virus strains.^{7,8} While the disruption of enveloped viruses by various classes of membrane-active agents (e.g., surfactants,^{9,10} alcohols,¹¹ photosensitizers¹²) has long been known to reduce infectivity *in vitro*, recent evidence points to

the *in vivo* and *ex vivo* efficacy of membrane-active antiviral peptides^{13,14} and thus steps closer to realize potential translational opportunities. Together, these inhibitory activities have led to the paradigm concept of lipid envelope antiviral disruption (LEAD),⁸ whereby an antiviral peptide-induced reduction in the extracellular concentration of infectious virus particles can lead to medically beneficial outcomes.

Within this scope, there is significant interest in understanding the basic design principles of membrane-active antiviral peptides, including how the structural properties of a peptide contribute to functional activity and the physicochemical basis of targeting selectivity.^{13,15} Since the lipid membrane surrounding enveloped viruses is derived from host cell membranes, the lipid compositions of viral

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membranes and infected cell membranes can be similar, and hence antiviral peptides need to have different mechanisms of targeting selectivity.¹⁶ Unlike the cationic antimicrobial peptides described above, two of the most widely studied, membrane-active antiviral peptides, the 27-mer AH and 18-mer CSA,¹⁷ have net charges of 0 and -2 , respectively. Interestingly, the AH and CSA peptides have partially overlapping amino acid sequences and are both derived from the N-terminus of the hepatitis C virus (HCV) nonstructural protein 5A (NSSA), which plays an important role in HCV genome replication, as illustrated in Figure 1.

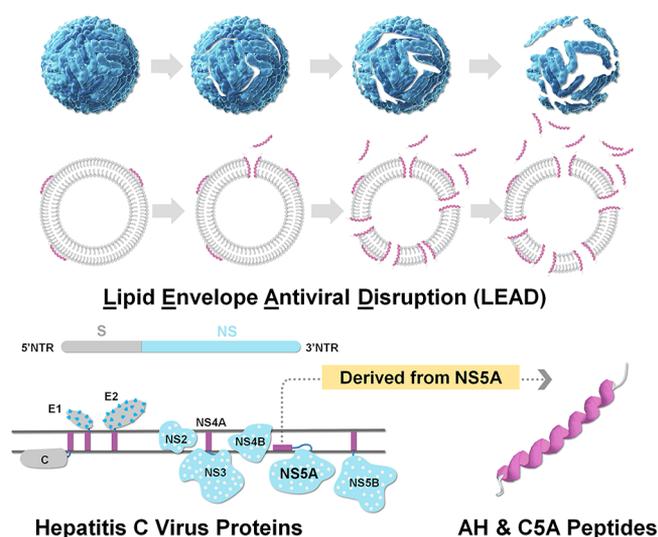


Figure 1. Schematic illustration of the LEAD concept and biological origin of the antiviral AH and CSA peptides, which correspond to the N-terminal amphipathic helix of the HCV NSSA protein.

Originally, the AH peptide was designed as a synthetic analogue to mimic the NSSA N-terminal amphipathic helix of an HCV genotype 1b strain (AH is an abbreviation for amphipathic, α -helix) for at least two reasons: (i) NSSA membrane association is necessary for HCV genome replication and dependent on the NSSA N-terminus,¹⁸ and (ii) it is difficult to express and purify full-length NSSA protein due to the amphipathic character of its N-terminus (it is often expressed without the N-terminus; see, e.g., refs 19–21). For these reasons, the AH peptide provided a useful model system to investigate the molecular determinants potentially influencing NSSA membrane association to model phospholipid and cell-derived membrane platforms.^{22–24} In the course of these early studies, it was serendipitously discovered that the AH peptide can rupture lipid vesicles below a certain diameter (~ 160 nm; see refs 25 and 26), and follow-up studies indicated that this membrane-disruptive behavior relates to membrane-curvature-dependent pore formation and resulting strain-induced membrane lysis once a critical density of pores is formed.^{25–28} In contrast to cationic antimicrobial peptides, the AH peptide has also been shown to disrupt lipid membranes largely independent of membrane surface charge.²⁹

While it remains unclear how the membrane-disruptive activity of the isolated AH peptide might relate to supporting HCV genome replication, it has been demonstrated that the AH peptide's membrane-disruptive activity also works *in vitro* against numerous, medically important enveloped viruses such as Zika and Dengue along with HCV itself. Using various

nanoscale characterization techniques, it has been shown that AH peptide treatment can disrupt virus particles in a similar manner to how it ruptures lipid vesicles.^{25,30} Importantly, due to its membrane-curvature-dependent mechanism, AH peptide exhibits high potency against enveloped viruses (half-maximal efficacy in the range 10 nM to 1 μ M depending on the virus and assay) while exhibiting low cytotoxicity against larger mammalian cells (half-maximal cytotoxicity around 50 μ M or higher depending on the cell type and assay).¹³ An engineered version of AH peptide that is composed of all D-amino acids was recently shown to work *in vivo* to therapeutically treat the lethal Zika virus infection in mice by reducing viral loads and blunting virus-related inflammation, and the peptide was also able to cross the blood–brain barrier.¹³ While it is primarily thought that the AH peptide works against extracellular virus particles, it has been speculated that it might also inhibit intracellular NSSA membrane association through competitive binding to cellular receptors.¹⁸

A few years after the AH peptide was first designed, the CSA peptide was discovered through a high-throughput screen that investigated the potential anti-HCV activity of 441 synthetic peptides that were derived from the genome of an HCV genotype 1a strain.¹⁷ Early work showed that the CSA peptide has potent antiviral activity against numerous enveloped viruses, including HCV, HIV, Dengue, and Herpes simplex (half-maximal efficacy in the range 500 nM to 5 μ M depending on the virus and assay).^{8,17,31,32} On the other hand, it was also reported that the CSA peptide exhibits greater hemolytic activity than the AH peptide,¹⁵ and thus, its ongoing development has focused on topical microbicide applications. To date, the CSA peptide has been shown to protect mice³³ and nonhuman primates¹⁴ against HIV transmission *ex vivo*. The current understanding is that the CSA peptide disrupts enveloped virus particles although it has also been shown to exhibit immune-stimulating activities, suggesting that it might be a useful adjuvant as well.³⁴ From a mechanistic perspective, an important distinction between the AH and CSA peptides is that the CSA peptide does not exhibit membrane curvature selectivity and ruptures lipid vesicles independently of vesicle size, which helps to explain why it has less targeting selectivity.^{8,15}

At the same time, such functional differences motivate a deeper investigation of the membrane-interaction profiles of these two antiviral peptides, including clarifying how functional differences might be related to variations in peptide conformational structure and membrane-induced peptide conformational changes. Indeed, comparison of the aligned AH and CSA sequences reveals that the two peptides are structurally similar at the primary amino acid level and most amino acid differences between the two aligned sequences are conservative substitutions, which is consistent with the fact that the NSSA AH is conserved across HCV genotypes, including the 1a and 1b genotypes that have $\sim 80\%$ genetic similarity overall.^{35,36} This similarity points to the likely importance of peptide conformational properties in driving membrane interactions and addressing this topic forms the focus of the present study.

Herein, using multiple surface-sensitive measurement techniques and computational simulations, we characterized the membrane-interaction profiles of AH and CSA peptides with model lipid membranes and sought to understand how peptide conformational changes drive these membrane interactions. Our experimental approach applied the nano-architectonics concept, whereby we selected two different

types of supported lipid membrane platforms, lipid vesicles with different degrees of membrane curvature, and a tethered lipid bilayer membrane (tBLM), in order to characterize the membrane–peptide interactions using different measurement techniques.^{37–40} To this end, circular dichroism spectroscopy experiments were conducted to determine peptide secondary structure in aqueous buffer and in the presence of lipid membrane environments, leading to insights into coil-to-helix folding and the thermodynamics of membrane–peptide interactions. These results were complemented by single-vesicle assay measurements to characterize the real-time interactions of the two peptides with lipid vesicles along with electrochemical impedance spectroscopy measurements to determine to what extent the peptide interactions cause membrane leakage in the tBLM configuration. Together, these experimental studies build a comprehensive picture of how AH and CSA peptides distinctly interact with lipid membranes, and additional computational simulations further provided insight into how these functional differences might relate to peptide conformational structure.

EXPERIMENTAL SECTION

Reagents. High-purity AH and CSA peptides (96% and 95%, respectively) were synthesized by AnaSpec Corporation (Fremont, CA, USA). The lyophilized peptide was weighed and solubilized in deionized water to prepare a highly concentrated stock concentration (2 mg/mL). The molar concentrations of the two peptides were determined by absorbance measurements at 280 nm wavelength, and a Boeco-S220 spectrophotometer (Boeco, Hamburg, Germany) was used for the experiments.⁴¹ The experimentally determined molar concentrations of AH and CSA peptides were 554 and 782 μM , respectively, which indicate that both peptides have a high solubility in water. The peptide stock aliquots were stored at $-20\text{ }^\circ\text{C}$ and diluted in buffer solution before the experiments. Lipids were obtained from Avanti Polar Lipids (Alabaster, AL, USA) and were supplied as a dispersion in a chloroform solution or in a lyophilized, powder form. Lipids included 1,2-dioleoyl-*sn*-glycero-3-phosphocholine (DOPC) for all experiments, and 1,2-distearoyl-*sn*-glycero-3-phosphoethanolamine-*N*-[biotinyl(polyethylene glycol)-2000] (ammonium salt) (DSPE-PEG(2000)biotin) and 1,2-dioleoyl-*sn*-glycero-3-phosphoethanolamine-*N*-(lissamine rhodamine B sulfonyl) (ammonium salt) (Rh-PE) were used for tethered vesicle experiments. All of the lipid reagents were mixed in chloroform to the desired molar ratio before vesicle preparation as described below. The aqueous buffer was 10 mM Tris [pH 7.5] buffer with 150 mM NaCl unless otherwise noted. All solutions were prepared using Milli-Q-treated water (MilliporeSigma, Burlington, MA, USA).

Circular Dichroism (CD) Spectroscopy. The secondary structure of peptides was characterized by CD spectroscopy experiments using a Chirascan spectropolarimeter (Model 420, AVIV Biomedical Inc., Lakewood, NJ, USA) with a 0.1 cm path length cuvette (Hellma GmbH, Müllheim, Germany). The test peptide concentration was fixed at 100 μM . Mixtures of peptide and lipid were mixed to the appropriate concentration (for example, a P:L ratio of 1:10 corresponds to 100 μM peptide and 1 mM lipid in the sample mixture) and then incubated for 30 min before experiment. The aqueous buffer used for all steps was 10 mM Tris [pH 7.5] buffer with 150 mM NaCl. The maximum lipid concentration was set at 2 mM, which is within the optimal range for data collection.⁴² Spectral data were collected in triplicate at 25 $^\circ\text{C}$, and the wavelength range was 190–260 nm with 1 nm bandwidth, a step size of 0.5 nm, and a time constant of 0.1 s. Background scans without peptide were recorded using the same conditions and subtracted accordingly. The final averaged spectra were calculated in mean residue molar ellipticity units ($[\theta] = \theta/10 \times c \times l$), where θ is the ellipticity, c is the molar concentration of peptide, and l is the path length in centimeter. The presented spectra were smoothed by the Savitzky–Golay method with

a polynomial order of 2. The fractional helicity (fH) of the peptides was calculated as follows

$$\text{fH} = \frac{[\theta]_{222}^{\text{Obs}} - 3000}{-36000 - 3000}$$

where $[\theta]_{222}^{\text{Obs}}$ is the molar ellipticity at 222 nm.^{15,43}

Tethered Vesicle Platform. Glass coverslips (ibidi GmbH, Martinsried, Germany) were cleaned and treated with oxygen plasma (model PDC-002, Harrick Plasma, Ithaca, NY, USA) for 30 s, before being assembled in a microfluidic chamber (sticky slide VI 0.4, ibidi GmbH). The coverslips were then coated with a mixture of 50/50 wt % poly(L-lysine)-grafted poly(ethylene glycol) (PLL-g-PEG) and poly(L-lysine)-grafted poly(ethylene glycol)-biotin (PLL-g-PEG-biotin) (SuSoS AG, Dübendorf, Switzerland) at a polymer mass concentration of 41.5 $\mu\text{g/mL}$. The incubation time was 30 min followed by a rinse in 10 mM Tris [pH 7.5, 150 mM NaCl] buffer. Then, the surface was incubated with 0.17 μM neutravidin for 5 min, followed by buffer washing. Biotinylated vesicles were next immobilized on the functionalized surface by biotin–neutravidin coupling at a lipid concentration of $\sim 0.03\text{ }\mu\text{M}$. Image analysis was performed using ImageJ software (National Institutes of Health, Bethesda, MD) and an in-house software program written in Python code, as previously described.¹⁵

Vesicle Preparation. For a tethered vesicle platform, small unilamellar vesicles (SUVs) were prepared by the extrusion method, as previously described.⁴⁴ Dissolved lipids (99.2 mol % DOPC, 0.7 mol % of Rh-PE and 0.1 mol % of DSPE-PEG(2000)biotin) in chloroform were first dried under a gentle stream of nitrogen gas at room temperature and were then stored under vacuum conditions overnight to remove any remaining chloroform by evaporation. Multilamellar vesicles were then generated by hydrating and vortexing the dried lipids in aqueous buffer to achieve a bulk lipid concentration of $\sim 2.5\text{ mM}$. The buffer contained 14.3 mM calcein dye (maximum excitation and emission wavelengths of 494 and 517 nm, respectively), which was thus incorporated into the lipid vesicles. After hydrating the lipid samples, the vesicles were subjected to seven cycles of freeze–thaw treatment to increase encapsulation efficiency and unilamellarity. Then, the lipid suspension was repeatedly passed through a polycarbonate filter a total of 11 times by using a MiniExtruder (Avanti Polar Lipids). Depending on the diameter of pores within the filter, the average diameter of extruded vesicles was 126 and 313 nm for 100 and 400 nm filters, respectively, as determined by dynamic light scattering (Figure S1). The as-prepared vesicle suspensions were diluted before the experiments to a lipid mass concentration of $\sim 0.25\text{ mM}$, and nonencapsulated calcein was removed by a Sephadex G-25 gel filtration column (GE Healthcare Life Sciences, Pittsburgh, PA, USA).⁴⁵

Epifluorescence Microscopy. Imaging experiments were conducted using a Nikon Eclipse Ti-E inverted microscope with a 60 \times oil-immersion objective (NA 1.49). The excitation source was a mercury-fiber illuminator C-HGFIE Intensilight (Nikon, Tokyo, Japan), and the light was passed through an alternating dichroic filter block (Ex 480/40, Em 535/50) or (Ex 545/30, Em 605/70) for imaging in the FITC and TRITC channels, respectively. An Andor iXon3 897 EMCCD camera (Andor Technology Ltd., Belfast, United Kingdom) was used to record images at a rate of 1 frame per 0.2–0.4 min in four neighboring spots. The experimental substrate was enclosed within a microfluidic chamber, and the liquid sample was introduced at a flow rate of 100 $\mu\text{L/min}$, as controlled by a peristaltic pump (model no. ISM833C, Ismatec, Wertheim, Germany). All measurements were conducted at room temperature ($\sim 25\text{ }^\circ\text{C}$).

Electrochemical Impedance Spectroscopy (EIS). The EIS measurement platform consisted of gold electrodes that were supplied by SDx Tethered Membranes (Roseville, New South Wales, Australia) and coated with 10 mol % tether (benzyl disulfide octoethylene glycol phytanyl) and 90 mol % spacer (hydroxyl terminated benzyl disulfide tetra-ethylene glycol) molecules dispersed in ethanol. The assembled tethaPlate (SDx Tethered Membranes) contained six flow cells with a 2.1 mm² electrode surface area per cartridge, and the

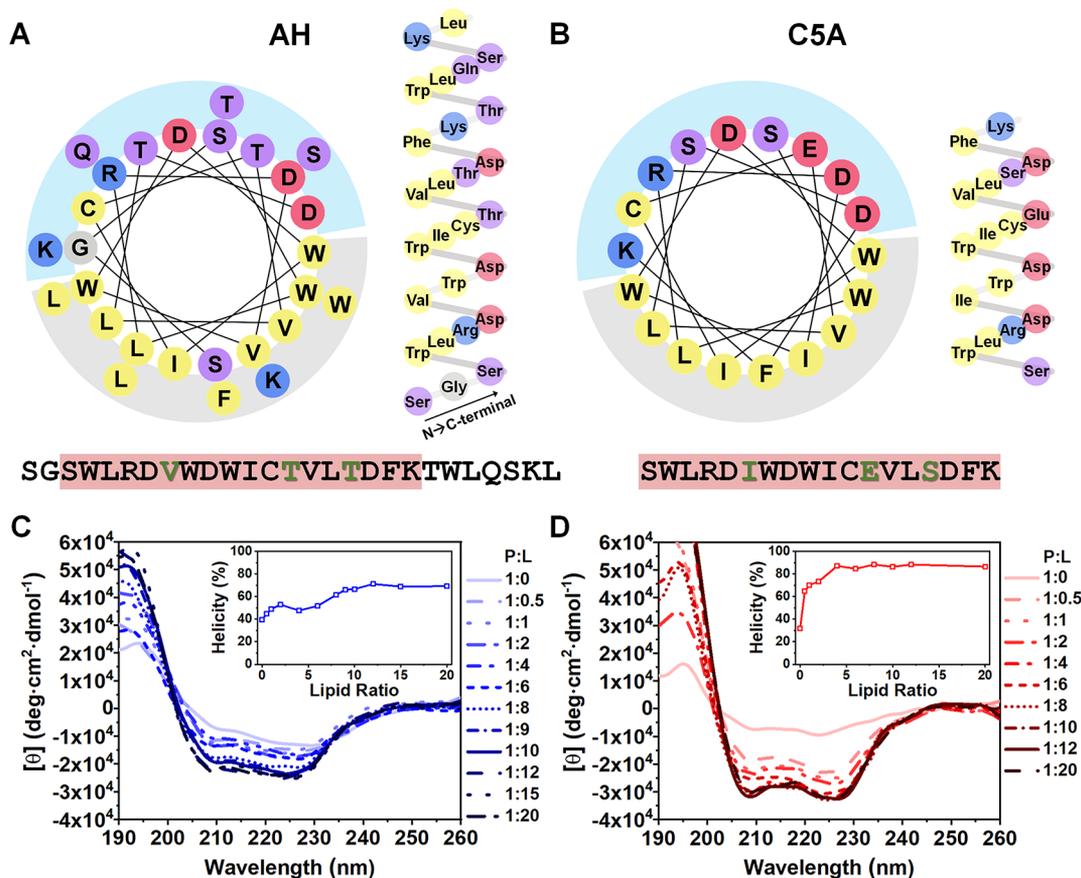


Figure 2. Secondary structure comparison of AH and C5A peptides. Helical wheel projection of (A) AH and (B) C5A amino acid sequences showing an amphipathic distribution of hydrophilic (purple), positively charged (blue), negatively charged (red), and hydrophobic (yellow) amino acid residues. Circular dichroism (CD) spectroscopy analysis of (C) AH and (D) C5A peptides in the presence of lipid concentrations (with fixed peptide concentration of 100 μM). Insets show the fractional helicity percentages of the two peptides as a function of normalized lipid concentration (defined as lipid concentration over peptide concentration; a value of 20 on the x -axis insets corresponds to 2000 μM lipid, for example). All presented spectra were obtained by averaging three technical replicates, and the helicity percentages were calculated on the basis of the mean ellipticity values at the 222 nm wavelength.

formation of tethered bilayer lipid membranes (tBLMs) on the sensor surfaces was initiated by the addition of 8 μL of 3 mM DOPC lipid in ethanol. After 2 min of incubation, the flow cell was rapidly rinsed with 10 mM Tris [pH 7.5, 150 mM NaCl] buffer (3 cycles \times 100 μL per cycle) in order to promote tBLM formation. Afterward, 100 μL of the appropriate test sample was injected into each well. TethaPod and tethaPatch units from SDx Tethered Membranes were used as the membrane conductance and capacitance reader and as the potentiostat connectivity unit, respectively. All measurements were collected under ambient room-temperature conditions ($\sim 25^\circ\text{C}$) and analyzed using the tethaQuick software program (SDx Tethered Membranes), as previously described.^{46,47}

Structural Modeling. The conformational properties of free peptide molecules in bulk solution were predicted by using the PEP-FOLD3 server, which is an online computational framework for *de novo* structural modeling.⁴⁸ The simulations were conducted without any preset data; i.e., the modeling was unbiased. Structural prediction occurs in three consecutive steps: (1) the peptide structure is described by a series of fragments of four amino acids, overlapping by three amino acids each; (2) conformations are generated on the basis of the prototype fragments followed by optional refinement with a Monte Carlo procedure; and (3) predicted conformations are ranked according to a scoring system.⁴⁹ The predicted conformational structures of free peptide monomers were visualized using the Swiss-PdbViewer program.⁵⁰

Statistical Analysis. Statistical analysis was performed using the Origin 2018 software package (OriginLab, Northampton, MA, USA). The number of vesicles analyzed per sample was typically >300. The

mean \pm standard deviation (sd) in histogram plots was determined by Gaussian fitting, and sd was defined as $\text{fwhm}/2.355$, where fwhm is the full-width-at-half-maximum.

RESULTS AND DISCUSSION

Secondary Structure Characterization. In Figure 2A,B, the helix net projections and amino acid sequences of the AH and C5A peptides are presented and show the amphipathic disposition of each peptide, as indicated by polar (blue) and nonpolar (gray) faces. A small disturbance in the AH peptide's amphipathicity is noted due to polar residues being introduced near each end of the nonpolar face (S1 and K26 in the N- and C-terminus, respectively). The 18-mer C5A peptide is related to the middle section of the 27-mer AH peptide with 3 amino acid differences that are highlighted in green. These differences represent conservative mutations that arise from substituting nonpolar with nonpolar (V8 to I6) or polar with polar (T14 to E12, T17 to S15) residues and relate to the different genotypic origins of the two peptides as discussed in the Introduction. As such, both peptides have similar near-neutral electrical properties, and the net charges of the AH and C5A peptides are 0 and -2 , respectively. Likewise, the hydrophobicity (H) of each peptide was comparatively high (0.69 and 0.74, see Table S1). Hence, the primary amino acid sequences of the AH and

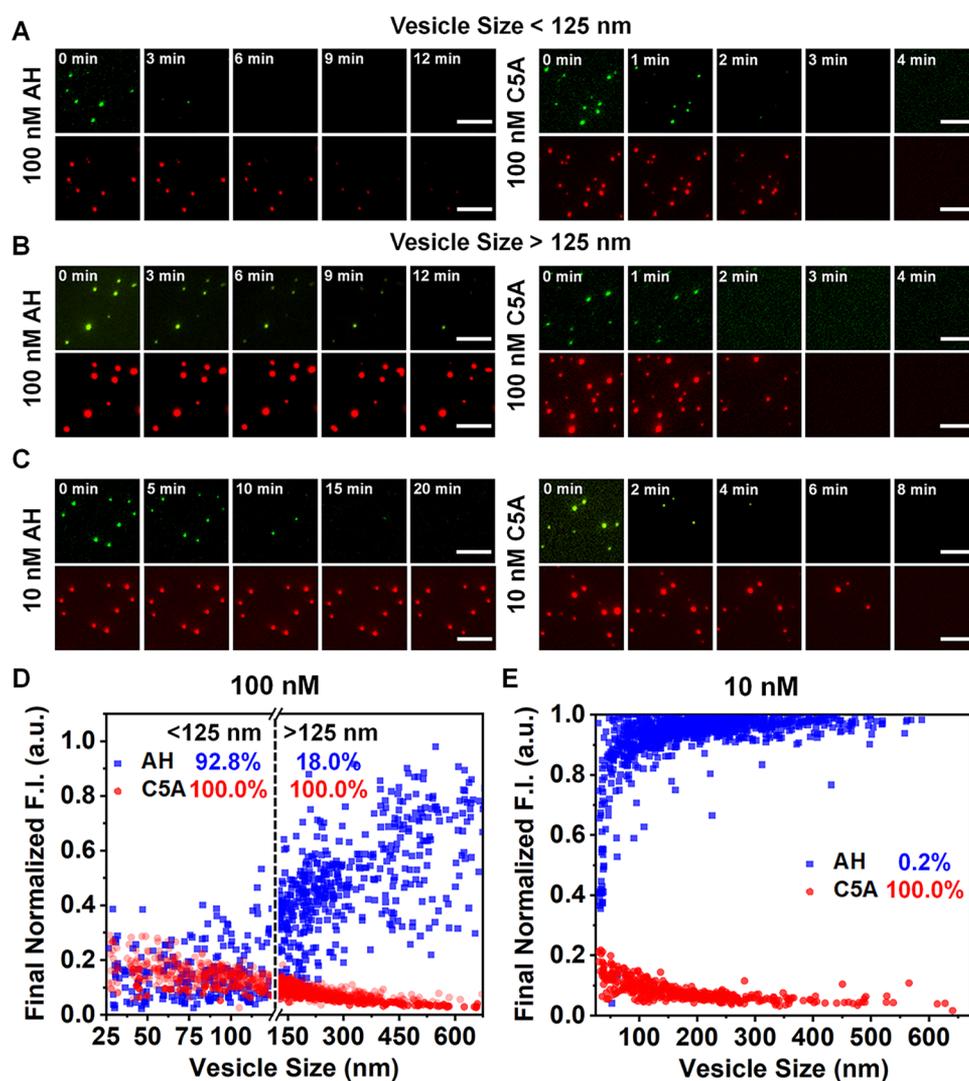


Figure 3. Distinct membrane-disruptive potencies of AH and C5A peptides. Time-lapsed fluorescence microscopy images of encapsulated calcein (green, top) and Rh-PE (red, bottom) dye probes for addition of 100 nM peptide to (A) small vesicles (<125 nm diameter) and (B) large vesicles (>125 nm diameter), or (C) 10 nM peptide to small vesicles (<125 nm diameter). Scale bars are 10 μm . (D, E) Corresponding single-vesicle data of peptide-induced vesicle rupture as a function of vesicle size, based on the rhodamine signal. Rupture is defined as when the final fluorescence intensity value of an individual vesicle was 30% or less of the original fluorescence intensity value of that vesicle. Each data point represents an individual vesicle.

C5A peptide are similar and endow the two peptides with similar physicochemical properties as well.

We next conducted lipid concentration-dependent CD spectroscopy experiments in order to measure the change in peptide secondary structure when peptide molecules partition from aqueous buffer solution into lipid membrane environments. This titration method also allowed us to identify how many lipid molecules are needed to interact with one peptide molecule in order to induce a stable α -helical conformation. Typically, amphipathic peptide molecules exhibit increased helicity in lipid membrane environments, and the magnitude of the helicity change provides insight into the free energy of membrane partitioning.^{51,52} In the free state, both peptides exhibited partial α -helical structures in aqueous buffer solution due to the competition between intramolecular peptide hydrogen bonds and intermolecular binding to solvent molecules (Figures 2C,D).⁵³ In lipid membrane environments, both peptides transitioned from a polar (solvent) to a nonpolar (lipid membrane) environment, and this transition coincided

with a change in the peptide secondary structure from a partial helix to an increased α -helical conformation that was visible in the CD spectra. Helix stabilization is a common element of membrane-peptide interactions, and the folded structure of a peptide molecule in a lipid membrane environment has a lower free energy (more stable) than the less-folded structure in bulk solution; the degree of helicity has also been shown to be related to the extent of permeabilization activity.^{54–56}

With increasing lipid concentration (in the range 0–2000 μM), the α -helical character of peptides was greater because more peptide molecules partitioned into the lipid membrane environment until saturation was reached. For the AH peptide, the helicity increased from 39% in aqueous buffer to 71% at high lipid concentrations, and helix stabilization was reached at a peptide-to-lipid (P:L) ratio of about 1:10. By contrast, for the C5A peptide, the helicity increased from 32% in aqueous buffer to 88% at high lipid concentrations, and helix stabilization was reached at a P:L ratio of about 1:6.

Together, the data indicate that the CSA peptide achieves more extensive helix stabilization in lipid membrane environments than AH peptide. CSA helix stabilization also occurs at a smaller P:L ratio, which is consistent with its greater membrane partitioning, and the combination of these findings supports that the CSA peptide has greater interfacial activity against lipid membranes and is more potent than the AH peptide at lower P:L ratios.

For both peptides, we also investigated the ratio of the magnitudes of the two bands at the 208 and 222 nm wavelength positions, respectively (defined as $[\theta]_{222}/[\theta]_{208}$). Depending on the value of the ratio, it can be ascertained whether the peptides exist in a coiled-coil structure ($\gg 1$) or as noninteracting helices (≈ 1).⁵⁷ When the AH peptide was in aqueous buffer, the ratio was 3.5, which indicates that peptide molecules exist as a coiled-coil structure (Figure S2). In the presence of increasing lipid concentrations, the $[\theta]_{222}/[\theta]_{208}$ ratio decreased to around 1.1, which is more consistent with the value expected for noninteracting, linear helices rather than coiled-coil structures. By contrast, the CSA peptide existed predominantly as noninteracting monomers in buffer solution, and the ratio reached ~ 1.0 in the presence of higher lipid concentrations. The results support that the AH peptide transitions from a coiled-coil helical structure to a noninteracting, helical structure in the presence of lipid membrane environments while the CSA peptide exists as noninteracting helices in both aqueous buffer and lipid membrane environments. Collectively, the findings support that AH and CSA peptides have distinct conformational changes associated with membrane partitioning, and such factors help to explain why the two peptides have different potencies.

Real-Time Tracking of Peptide-Induced Vesicle Rupture. To directly track membrane–peptide interactions, we conducted time-lapsed fluorescence microscopy experiments in a microfluidic chamber whereby the peptide-mediated rupture of individual, tethered lipid vesicles on a functionalized glass surface was monitored in real-time. The vesicles were labeled with two fluorescence probes: a water-soluble calcein dye (green) was encapsulated in the vesicle interior, and a rhodamine dye-labeled phospholipid Rh-PE (red) was incorporated into the vesicle bilayer in order to track membrane permeabilization and lysis, respectively. The two types of membrane-interaction events were detected on the single-vesicle level on the basis of monitoring decreases in the intensity of the two fluorescence signals per individual vesicle, and over 300 vesicles could be monitored simultaneously per experiment.

For small vesicles (<125 nm diameter), 100 nM AH peptide ruptured nearly all vesicles within around 12 min while 100 nM CSA peptide ruptured all vesicles within only 3 min (Figure 3A). By contrast, for larger vesicles (>125 nm diameter), 100 nM AH peptide only ruptured a minor fraction of vesicles and most tethered vesicles remained intact (Figure 3B). On the other hand, 100 nM CSA peptide was still active against larger vesicles and ruptured them within 3 min. At 10 nM peptide concentration, the AH peptide lost membrane-disruptive activity against even small vesicles (<125 nm diameter) whereas the CSA peptide maintained similar activity levels and caused vesicle rupture within ~ 10 min (Figure 3C).

We also plotted the final intensity values of the rhodamine signal per individual vesicle versus each individual vesicle's size in order to quantitatively determine what fraction of vesicles was “ruptured” in different size categories.⁵⁸ For 100 nM AH

peptide, 92.8% of small vesicles (<125 nm diameter) were ruptured while only 18.0% of larger vesicles were ruptured (>125 nm diameter) (Figure 3D). This vesicle size-dependent trend contrasted with the results obtained with 100 nM CSA peptide, in which case the peptide ruptured 100% of vesicles across the tested range of vesicle sizes. Strikingly, 10 nM AH peptide displayed negligible activity (0.2% rupture efficiency) against small vesicles (<125 nm diameter) whereas 10 nM CSA peptide still exhibited 100% rupture efficiency (Figure 3E).

Quantitative analysis of the vesicle rupture kinetics further confirmed that the CSA peptide is more potent (ruptures lipid vesicles at a 10-fold lower bulk peptide concentration than AH peptide) and causes membrane permeation and lysis on quicker time scales than AH peptide. Treatment with 100 nM AH peptide caused rapid permeabilization of small vesicles in around 2 min while membrane lysis typically occurred within 5 min (Figure 4A). However, 100 nM AH peptide caused much

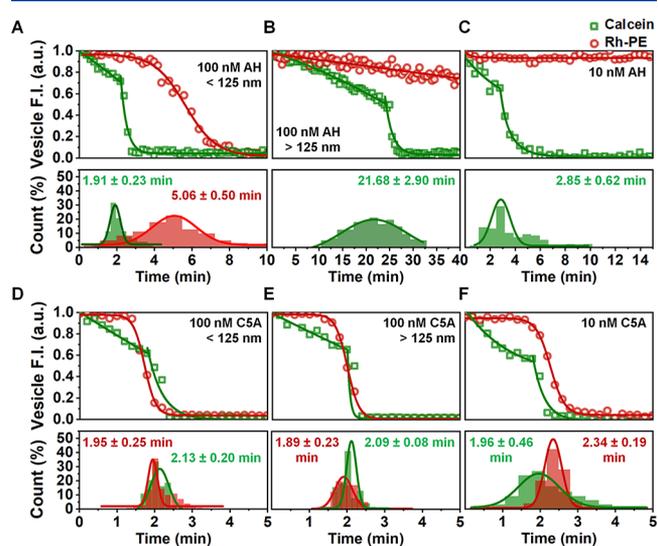


Figure 4. Change in normalized fluorescence intensity of individual, tethered vesicles as a result of membrane–peptide interactions. (A–F) Top: Representative profiles of single-vesicle rupture kinetics for AH and CSA peptides at different peptide concentrations and for different vesicle size ranges. Green and red colors denote calcein and rhodamine signals, respectively. Bottom: Corresponding histogram statistics of membrane permeabilization (green) and membrane lysis (red) time scales, where applicable.

slower permeabilization of larger vesicles without inducing membrane lysis (Figure 4B). Likewise, 10 nM AH peptide caused membrane permeabilization of smaller vesicles but did not cause membrane lysis (Figure 4C). As such, AH peptide-mediated pore formation is selective for small vesicles while a relatively high bulk peptide concentration is required to cause membrane lysis. By contrast, 100 nM CSA peptide is able to rapidly permeabilize and lyse both small and large vesicles on similar time scales of around 2 min (Figures 4D,E). Nearly identical results were obtained when 10 nM CSA peptide was added to small vesicles as well (Figure 4F). Thus, CSA peptide exhibits more potent, indiscriminate activity against lipid vesicles, as indicated by vesicle-independent rupture activity and rapid kinetics. Of particular note, the time scales of membrane permeabilization and membrane lysis were nearly identical for CSA peptide, suggesting that its membrane-interaction profile is distinct from that of AH peptide which

induces membrane lysis *after* a critical density of pores is formed. On the other hand, AH peptide exhibits vesicle size-dependent rupture activity, and the rate of membrane permeabilization is vesicle size-dependent as well, which is consistent with its known membrane curvature-sensitive, pore-forming behavior.

Peptide-Induced Membrane Leakage. To further investigate the effects of AH and CSA peptide treatment on membrane permeability, we conducted electrochemical impedance spectroscopy (EIS) measurements using a tethered bilayer lipid membrane (tBLM) platform. A schematic illustration of the measurement setup is presented in Figure 5A, and the experimental objective was to detect membrane

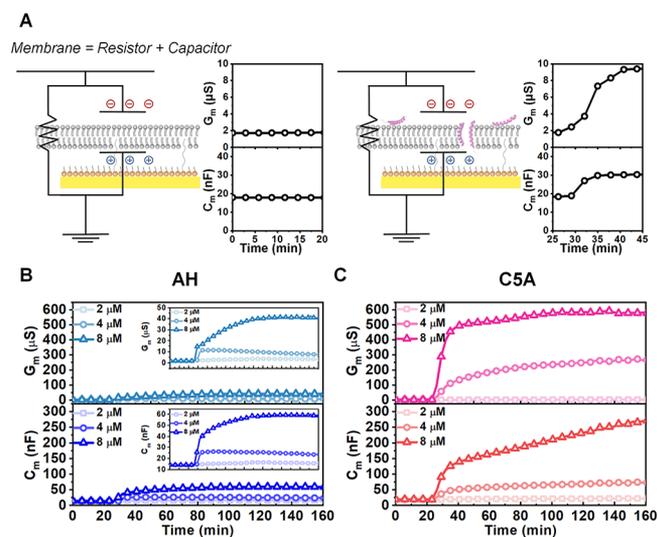


Figure 5. Electrochemical impedance spectroscopy analysis of membrane–peptide interactions involving AH and CSA peptides. (A) Schematic illustration of measurement setup. Peptide concentration-dependent changes in G_m (top) and C_m (bottom) signals as a function of time for (B) AH and (C) CSA peptides, respectively. Peptides were added at approximately $t = 20$ min.

permeabilization based on ion leakage across the lipid bilayer. The setup measures the electrical conductance (G_m) across the tBLM, which is the inverse of its electrical resistance and expressed as $G_m = I/V$, where I and V are the current and voltage, respectively. The setup also measures the capacitance (C_m) of the tBLM, which signifies the magnitude of charged ions that are stored across the lipid bilayer and is expressed as $C_m = q/V$, where q and V are the charge and voltage, respectively. When membrane leakage occurs, the G_m and C_m values both increase (larger increases indicate greater membrane leakage) and a well-sealed tBLM is important for baseline stabilization. In our experiments, the tBLM was formed by rapid solvent exchange, and the initial G_m and C_m values were less than $2 \mu S$ and around 15–25 nF, respectively, which indicated good sealing properties.⁵⁹

Figure 5B,C presents the concentration-dependent measurements results for AH and CSA peptides, respectively. Both peptides showed concentration-dependent membrane leakage activity, with distinct differences in the level of membrane permeabilization. At the highest test concentration, 8 μM AH peptide caused stable G_m and C_m shifts of around 40 μS and 60 nF, respectively. In marked contrast, 8 μM CSA peptide caused appreciably larger G_m and C_m shifts of around 550 μS and 270 nF, respectively, and the C_m shift continued to increase over

the measurement time period. The distinct membrane–interaction profiles are consistent with the AH peptide causing membrane thinning⁶⁰ while the CSA peptide exhibited more indiscriminate membrane solubilization, which bore resemblance to the behavior of sodium dodecyl sulfate (SDS) surfactant in terms of the extent of membrane disruption, albeit with slower kinetics (Figure S3). Together with the foregoing experimental data, the EIS results indicate that AH and CSA peptide have distinct membrane–interaction profiles that arise from unique conformational properties.

Structural Modeling of Peptide Conformations. To further evaluate the conformational properties of the two peptides, we modeled their secondary structures by performing computational simulations with the PEP-FOLD online server.^{61,62} Figure 6A presents the simulated probabilities that the different amino positions are in helix, extended, or random conformations, as indicated by red, blue, and green colors, respectively. In general, it was predicted that both ends of the AH peptide have a high probability of helical conformation while it was predicted that the middle region of AH peptide (W9–18D) is likely nonhelical. On the other hand, the simulation results predicted that the CSA peptide has more uniform helical character throughout its amino acid sequence. On the basis of these predictions, the highest-ranking conformers of AH peptide mainly consisted of two helical regions that are separated by a nonhelical hinge in the middle (Figure 6B; see also Figures S4 and S5). This finding is consistent with the CD spectroscopy results indicating that the AH peptide likely has a coiled-coil structure, whereby the two helical regions of a peptide monomer interact and stabilize one another while the nonhelical hinge supports conformational freedom to minimize the peptide free energy. By contrast, the highest-ranking conformers of the CSA peptide have linear helix structures, which also agrees well with the CD spectroscopy results.

Taking into account the experimental and simulation results, our findings support that AH and CSA peptides exhibit distinct conformational properties that influence their membrane-disruptive activities, as presented in Figure 6C. While the AH peptide exhibits some degree of conformational flexibility in bulk solution, the simulation results indicate that it likely has two helical segments bifurcated by a flexible hinge, and this conformational arrangement offers insight into its empirically observed membrane curvature sensing properties. In high-curvature lipid membranes, there can be hydrophobic defects between neighboring lipid molecules,⁶³ and it appears that the AH peptide conformation is well-suited for fitting within these defects.

On the other hand, free CSA monomers exhibit lower helical character in the bulk solution while demonstrating greater membrane partitioning, as indicated by helix stabilization in the presence of lower total lipid concentrations in the CD spectroscopy experiments. As a result, the CSA peptide undergoes more extensive conformational changes, marked by a significant increase in helical character, upon membrane partitioning, which drives higher levels of interfacial activity.^{56,64}

As such, the CSA peptide is less discriminate of lipid membrane properties, and there is a larger thermodynamic driving force to favor membrane partitioning. This finding explains why the CSA peptide is more potent than the AH peptide and interacts with lipid membrane via a distinct mechanism that is more akin to membrane solubilization than

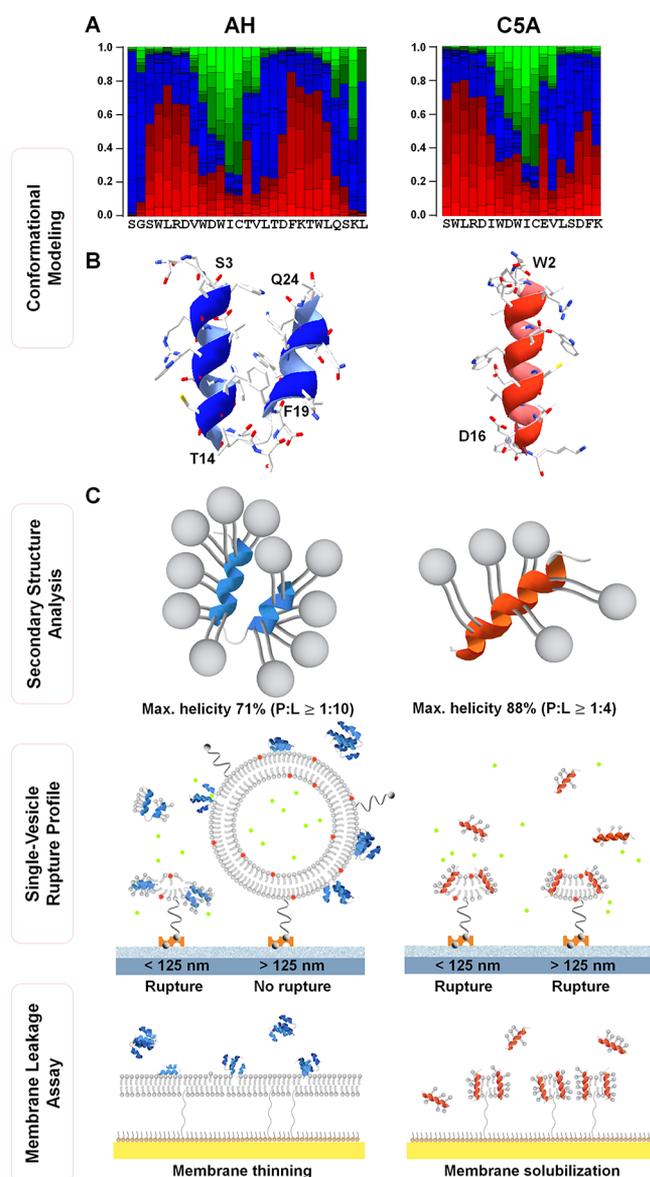


Figure 6. Structural modeling of AH and CSA peptides and implications for targeting selectivity. (A) Simulated probabilities of helical (red), extended (blue), and random (green) conformations are shown for each amino acid position in the peptide sequences. (B) Highest-ranking conformers of AH and CSA peptides are presented, pointing to distinct conformational properties of the two peptides in bulk solution. (C) Schematic illustration of how AH and CSA peptides interact with lipid membranes. Elements are not drawn to scale.

the curvature-sensitive pore formation exhibited by the AH peptide. Collectively, our findings reveal how the conformational properties of AH and CSA peptides influence structure–function relationships and help to explain the distinct membrane–interaction profiles of the two antiviral peptides. From a structural perspective, such information also provides a starting point to engineer antiviral peptides with desirable conformational properties and to screen candidate amino acid sequences using *in silico* methods and surface-sensitive measurement strategies. Looking forward, it will also be useful to apply this integrative approach across experiments and computational simulations to characterize the interactions of membrane-active peptides and other compounds with complex

lipid membrane models and in conjunction with other surface-sensitive measurement techniques as well.^{65,66}

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: [10.1021/acs.langmuir.9b01052](https://doi.org/10.1021/acs.langmuir.9b01052).

Additional information is provided about dynamic light scattering experiments (Figure S1), circular dichroism spectroscopy experiments and data analysis (Figure S2), electrochemical impedance spectroscopy experiments (Figure S3), and structural modeling simulations (Figures S4 and S5) (PDF)

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Notes

The authors declare the following competing financial interest(s): N.-J.C. is a co-inventor on US patent no. 8,728,793 that is related to the application of AH peptide molecules for antiviral therapy. The other authors declare no competing interests.

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